## A NEW FLAVONE METHYL ETHER FROM HELICTERES ISORA

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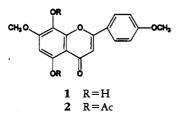
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ABSTRACT.—A new flavone, 5,8-dihydroxy-7,4'-dimethoxyflavone [1], has been isolated, along with trifolin and hibifolin, from the fresh leaves of *Helicteres isora*.

In a continuation of our study of the flavonoids of the Sterculiaceae (1-4), we report herein the isolation and characterization of a new flavone methyl ether, 7.4'-di-O-methylisoscutellarein (5.8dihydroxy-7,4'-dimethoxyflavone) [1] along with kaempferol-3-0-galactoside (trifolin) and herbacetin-8-0-glucuronide (hibifolin) from the fresh leaves of Helicteres isora L. This is the first report of the occurrence of 7,4'-di-O-methylisoscutellarein in nature. Although 8-oxygenated flavonols (3) and 6-oxygenated flavones (1,2) have been reported earlier from plants of the Sterculiaceae, this is the first record of the occurrence of an 8oxygenated flavone in this family.

The EtOH extract residue of fresh leaves was fractionated using petroleum ether (bp 60–80°), Et<sub>2</sub>O, EtOAc, and methyl ethyl ketone, successively. The Et<sub>2</sub>O and EtOAc fractions, on concentration, afforded trifolin and hibifolin, respectively, identified by direct comparison (mmp and co-paper chromatography) with authentic samples (3).

The methyl ethyl ketone fraction on concentration afforded a bright yellow compound [1] mp 261° (dec), which gave a gossypetone reaction. The isolate was purple under uv and uv/NH<sub>3</sub>, and gave an olive-green color with Fe<sup>3+</sup>. The eims of 1 exhibited a peak at m/z 314,



comparable with a molecular formula of  $C_{17}H_{14}O_6$ . Compound 1 showed uv absorptions characteristic of a flavone  $\{\lambda\}$ max (MeOH) 281, 305 nm], and its uv spectrum with diagnostic shift reagents revealed the absence of free C-7 and C-4' hydroxyls. The <sup>1</sup>H-nmr spectrum showed the presence of six aromatic protons as two doublets (H-2', H-6' and H-3', H-5') and two singlets (H-3, H-6, or H-8), and also revealed the presence of two methoxyl groups. On acetylation (Ac<sub>2</sub>O/  $HClO_4$ , 1 afforded a diacetate [2], and on demethylation (Ac<sub>2</sub>O/HI), it yielded 5,7,8,4'-tetrahydroxyflavone (isoscutellarein) (5). The presence of a OMe-4' group was also evident by the fragment ion  $\mathbf{B}_2^+$  at m/z 135 in the ms of the diacetate [2]. Based on the above data (color reactions, uv, and <sup>1</sup>H-nmr), **1** was concluded to be 5,8-dihydroxy-7,4'-(7.4'-di-0dimethoxyflavone methylisoscutellarein).

The structure of **1** was further supported by the <sup>13</sup>C-nmr spectrum of its diacetate [**2**]. The absence of an aromatic methine carbon signal in the range 90–96 ppm indicated that the C-8 was substituted, and the presence of acetoxyls at C-5 and C-8 was supported by their chemical shifts ( $\delta$  169.58 and 167.88) (6).

## EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Mps are uncorrected. Ir spectra were recorded in KBr. <sup>1</sup>H- (400 MHz) and <sup>13</sup>C- (100 MHz) nmr spectra were recorded on a JEOL GSX 400 spectrometer, using TMS as internal standard. Mass spectra (eims) were recorded on a JEOL-DX 300 spectrometer.

PLANT MATERIAL.—Leaves of Helicteres

isora (7) were collected from Alagar Hills, Madurai, India in the autumn of 1992, and identified by Prof. T. Sriganasen (Department of Botany, Madurai College, Madurai, India) and a voucher specimen was deposited at Madurai College Herbarium, Madurai, India.

EXTRACTION AND ISOLATION.-Fresh leaves (1 kg) were extracted with hot 90% EtOH under reflux and the subsequent concentrate was partitioned using petroleum ether, Et<sub>2</sub>O, EtOAc, and methyl ethyl ketone, successively. Concentration of the Et<sub>2</sub>O fraction afforded trifolin (50 mg), and hibifolin (30 mg) was obtained from the EtOAc fraction. The residue from the methyl ethyl ketone fraction was dissolved in a minimum volume of MeOH and refrigerated for 24 h to yield a greenish yellow solid, which on crystallization (MeOH) yielded a bright yellow compound [1] (120 mg); mp 261° (dec), purple under uv and uv/NH<sub>3</sub>, olive-green with  $Fe^{3+}$ ; uv (MeOH)  $\lambda$  max (log  $\epsilon$ ) 281 (4.10), 305 (4.41) nm; (+NaOAc/H<sub>3</sub>BO<sub>3</sub>) 232, 307 nm; (+AlCl<sub>3</sub>/HCl) 287, 320, 347, 426 nm; (+NaOMe) 269, 321 (sh) nm; ir (KBr) v max 3200 (br), 1650 (C=O), 1595, 1420, 830 (1,4disubstituted benzene), 750 cm<sup>-1</sup>; <sup>1</sup>H nmr (DMSO $d_{6}$ , 400 MHz)  $\delta$  3.87 (3H, s, OMe), 3.91 (3H, s, OMe), 6.56 (1H, s, H-3), 6.88 (1H, s, H-6), 7.13 (2H, d, J=8 Hz, H-3', H-5'), 8.15 (2H, d, J=8 Hz, H-2', H-6'), 8.91 (1H, s, OH-8), 12.44 (1H, s, OH-5); eims m/z [M]<sup>+</sup> 314 (97), 313 (84), 284 (35), 281 (100), 135 (10), 132 (5); anal., calcd for C<sub>17</sub>H<sub>14</sub>O<sub>6</sub>, C 64.97, H 4.46, found C 63.78, H 4.18.

ACETYLATION OF **1**.—Treatment of **1** (60 mg) with  $Ac_2O(2.0 \text{ ml})$  and  $HClO_4(0.2 \text{ ml}, 70\%)$  at room temperature overnight gave a crystalline diacetate [**2**] (50 mg), mp 236°; <sup>1</sup>H nmr (CDCl<sub>3</sub>, 60 MHz)  $\delta$  2.41 (6H, s, 2×OAc), 3.85 (3H, s, OMe), 3.90 (3H, s, OMe), 6.47 (1H, s, H-3), 6.68

(1H, s, H-6), 7.00 (2H, d, J=8 Hz, H-3', H-5'), 7.70 (2H, d, J=8 Hz, H-2', H-6'); eims m/z [M]<sup>+</sup> 398 (5), 314 (100), 312 (4), 135 (10); <sup>13</sup>C nmr (CDCl<sub>3</sub>, 100 MHz)  $\delta$  176.24 (C-4), 169.58, 167.88 (acetoxyl carbonyls), 162.51 (C-2), 162.03 (C-4'), 155.05 (C-7), 150.01 (C-5), 147.63 (C-9), 127.83 (C-2', C-6'), 123.72 (C-8, C-1'), 114.59 (C-3', C-5'), 111.19 (C-10), 106.95 (C-6), 104.29 (C-3), 56.64, 55.53 (methoxyl), 21.23, 20.30 (acetoxyl methyls).

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